Density dependence of the spur lifetime and the "escape" hydrated electron yield in the low-LET radiolysis of supercritical water at 400 $^{\circ}$ C

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Summary

The "spur lifetime" (τ_s) in the low-LET radiolysis of supercritical water (SCW) at 400 °C has been determined as a function of density by using a simple model of energy deposition initially in spurs, followed by the random diffusion of the species formed until spur expansion is complete. The values of τ_s are found to decrease from $\sim 5.0 \times 10^{-6}$ to $\sim 5.0 \times 10^{-8}$ s over the density range ~ 0.15 to 0.6 g/cm³. Using Monte-Carlo simulations, our calculated density dependence of the "escape" hydrated electron (e^-_{aq}) yield (i.e., at time τ_s) reproduces fairly well Bartels and co-workers' scavenged e^-_{aq} yield data, strongly suggesting that these data may have been measured at time close to τ_s .

1. Introduction

The radiolysis of water in the supercritical regime ($t_c = 373.95$ °C, $p_c = 22.06$ MPa and $\rho_c =$ 0.322 g/cm³) is an important subject in nuclear science and technology, especially for research and development of the next generation of water-cooled reactors (Gen-IV) [1]. Gen-IV supercritical water-cooled reactors (SCWRs) are promising advanced nuclear systems because of their high thermal efficiency (i.e., ~45% vs. 33% for current light water reactors) and simpler reactor designs. A key challenge in controlling the water chemistry of an SCWR will be the ability to mitigate the effects of water radiolysis [2]. The formation of oxidizing species, such as OH, H₂O₂, O₂, and O₂. (or HO₂, depending on pH), by water radiolysis can lead to corrosion and degradation of materials which can easily occur under the extreme irradiation conditions proposed for Gen-IV SCWRs (i.e., pressure of 25 MPa, temperatures as high as 625 °C, and intense flux of ionizing radiations – fast neutrons, γ -rays, recoil protons and heavy ions). However, the radiation-induced chemistry (especially, radiolytic yields and reaction rates) in SCW remains largely unknown [3]. Furthermore, there are still some apparent discrepancies on the available experimental data. For example, a strong difference has been observed in "scavenging" experiments between the e-aq yields in low-density SCW measured by Bartels and co-workers [4,5] in Notre Dame, U.S.A. (1.5 molec./100 eV at a density near 0.2 g/cm³) and those reported by Lin et al. [6,7] in Tokai, Japan (5.3 molec./100 eV at the same density). The hypothesis has been made that the N₂O and SF₆ scavenging yield data of Bartels and co-workers

¹ Throughout this paper, radiation chemical yields are quoted in units of molecules per 100 eV (abbreviated as "molec./100 eV"), as G_X or g(X) for primary (or "escape") yields and G(X) for experimentally measured yields. For conversion into SI units (mol/J): 1 molec./100 eV \approx 0.10364 μ mol/J.

and the MV^{2+} and 4,4'-bipyridyl scavenging results from Lin et al. may have been measured at different times.

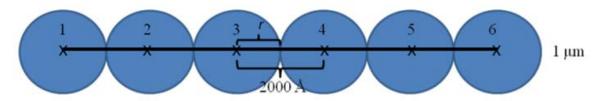
The aims of this work are two-fold; (1) to calculate the "spur lifetime" (τ_s) in the low-linear energy transfer (LET) radiolysis of SCW (H₂O) at 400 °C as a function of water density and (2) to better understand the differences that have been observed experimentally in the density dependence of the e^-_{aq} yield in irradiated SCW at 400 °C.

2. Procedure used to estimate the "spur lifetime" τ_s

To calculate τ_s as a function of density at 400 °C, we developed a simple procedure, *independent* of the presence of any scavengers, and based primarily on the value of the LET of the studied radiation and the random (Brownian) motion of the various diffusing species [8] formed in the radiolysis of water. This method allowed us to estimate a "mean" lifetime of the spur by simply averaging all individual $(\tau_s)_i$ values corresponding to each of the (main) radiolytic species, namely, e^-_{aq} , H^{\bullet} , H^{\bullet} , H^{\bullet} , and H^{\bullet} , and H

$$r^2 = 6 D_i (\tau_s)_i$$
 (1)

where r is the radius of the spur (assumed to be spherical) at *complete spur overlap* (i.e., when the spurs on expanding by diffusion will touch each other) and D_i is the diffusion coefficient of the species i in the medium. As illustrated below, r can be calculated from the LET of the considered radiation and the average energy being deposited in a spur.



- For ~300-MeV incident protons in liquid water at 25 °C (mimicking an irradiation with 60 Co γ -rays or fast electrons): LET ~ 0.3 keV/ μ m.
- Average energy loss in a spur: \sim 47 eV (as obtained from our Monte-Carlo track-structure calculations [10]), that is, 6 interactions (or "spurs") in a \sim 1- μ m proton track length.
- Distance between two adjacent spurs: 1 μ m/5 $\approx 2000 \text{ Å} \rightarrow r \approx 2000 \text{ Å/2} = 1000 \text{ Å}$.
- Spur lifetime for the species i (minimum time required before this species can be regarded as being homogeneously distributed), inferred from Eq. (1): $(\tau_s)_i = r^2/6 D_i$ ($i = e^-_{aq}$, H^{\bullet} , OH, H_2 , H_2O_2 , H^+ , OH⁻).

- "Average" spur lifetime:
$$\tau_s = \frac{\sum_i (\tau_s)_i G_i}{\sum_i G_i}$$

where G_i is the primary (or "escape") yield of the species i [11] (as different radiolytic species

have different primary yields, the "average" spur lifetime is obtained by summing over all the individual $(\tau_s)_i$ values corresponding to each of the species present in the spur after appropriate weighting is made according to their respective yields G_i).

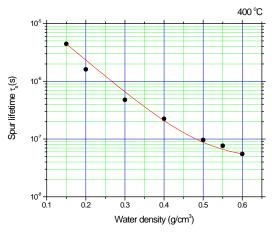
This calculation was repeated here at 400 °C at different densities between ~0.15 and 0.6 g/cm³.

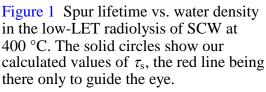
3. Monte-Carlo simulations

The stochastic modeling of the low-LET radiolysis of SCW (H_2O) at 400 °C was performed using our Monte-Carlo code IONLYS-IRT that simulates, in a 3D geometrical environment, the initial production of the various radiolytic species and the subsequent diffusion and chemical reactions of these species. The detailed description of the code is given in ref. [12]. This code has, however, been updated here using the radiolysis database (including rate constants, reaction mechanisms, and G-values) recently compiled by Elliot and Bartels [13]. The re-evaluation of certain parameters (e.g., the thermalization distance of subexcitation electrons) that intervene in the early physicochemical stage of the radiolysis has also been incorporated [14].

4. Results and discussion

Our calculated values of τ_s as a function of water density at 400 °C are shown in Fig. 1. They are found to decrease from $\sim 5.0 \times 10^{-6}$ s at 0.15 g/cm³ to $\sim 5.0 \times 10^{-8}$ s at 0.6 g/cm³. Such a result





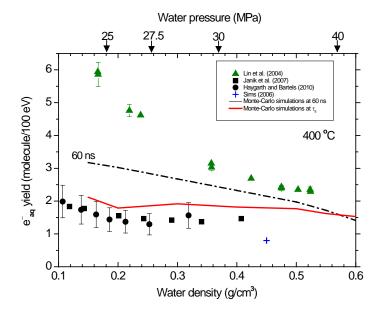


Figure 2 e^-_{aq} yield (in molec./100 eV) vs. water density in the low-LET radiolysis of SCW (H₂O) at 400 °C. The dash-dot line shows our calculated primary (or "escape") e^-_{aq} yields assuming that $\tau_s = 60$ ns is constant as a function of the density [15,16]. The solid (red) line shows our calculated yields using the spur lifetimes shown in Fig. 1. Symbols are experimental data [4-7,17] shown for comparison.

reflects the fact that the distance between two adjacent spurs becomes shorter at higher density, and therefore all the species inside the spurs will require a *shorter* time for homogeneity to become established.

As we can see from Fig. 2, our calculated density dependence of the "escape" e^-_{aq} yields (that is, the yields at times τ_s) – using the spur expansion times shown in Fig. 1 – reproduces fairly well the experimental data of Bartels and co-workers [4,5]. These results suggest strongly that the scavenged e^-_{aq} yields reported by these authors may have been measured at times that are close to the lifetimes of the spurs. In that respect, the present calculations corroborate very well the recent discussion of Haygarth and Bartels [5] who suggested that their "experiments probe something close to an escape yield". In contrast, scavenging in Lin et al.'s experiments [6,7] could have occurred at earlier time, of the order of a few nanoseconds at high water density (~0.5-0.6 g/cm³) but probably of the order of several tens of picoseconds at low water density (~0.15-0.2 g/cm³) as recently suggested by Muroya et al. [18] who first reported the kinetics of the decay of e^-_{aq} in SCW (D₂O) at 400 °C, using picosecond time-resolved pulse radiolysis experiments in the range ~60 ps to 6 ns.

5. Conclusion

In this work, we calculated the spur lifetime τ_s or, in other words, the time required for the changeover from nonhomogeneous spur kinetics to homogeneous kinetics in the bulk solution in SCW at 400 °C over the density range ~0.15 to 0.6 g/cm³. The results show that τ_s decreases with density from ~5.0 × 10⁻⁶ s at 0.15 g/cm³ to ~5.0 × 10⁻⁸ s at 0.6 g/cm³. The stochastic modeling of the radiolysis of SCW at 400 °C using Monte-Carlo simulations and the values thus found for τ_s showed that our calculated density dependence of the "escape" e^-_{aq} yield reproduced fairly well Bartels and co-workers' scavenged e^-_{aq} yield data, suggesting strongly that these data may have been measured at times close to τ_s . The present results thus shed new light on the apparent discrepancy that is observed in the various experimental density dependences reported so far for the e^-_{aq} yield in irradiated SCW at 400 °C.

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6. References

- [1] D. Guzonas, F. Brosseau, P. Tremaine, J. Meesungnoen, and J.-P. Jay-Gerin, "Water chemistry in a supercritical water-cooled pressure tube reactor", *Nucl. Technol.*, Vol. 179, Aug. 2012, in press.
- [2] D.M. Bartels, M. Anderson, P. Wilson, T. Allen, and K. Sridharan, "Supercritical water radiolysis chemistry. Supercritical water corrosion", 2006, Idaho National Laboratory Web site: http://nuclear.inl.gov/deliverables/docs/uwnd_scw_level_ii_sep_2006_v3.pdf
- [3] D.A. Guzonas, C.R. Stuart, J.-P. Jay-Gerin, and J. Meesungnoen, "Testing requirements for SCWR radiolysis", Report AECL 153-127160-Rept-001, Atomic Energy of Canada Ltd., Chalk River, Ontario, 2010.

- [4] D. Janik, I. Janik, and D.M. Bartels, "Neutron and β/γ radiolysis of water up to supercritical conditions. 1. β/γ yields for H₂, H^{*} atom, and hydrated electron", *J. Phys. Chem. A*, Vol. 111, 2007, pp. 7777-7786.
- [5] K. Haygarth and D.M. Bartels, "Neutron and β/γ radiolysis of water up to supercritical conditions. 2. SF₆ as a scavenger for hydrated electron", *J. Phys. Chem. A*, Vol. 114, 2010, pp. 7479-7484.
- [6] M. Lin, Y. Katsumura, Y. Muroya, H. He, G. Wu, Z. Han, T. Miyazaki, and H. Kudo, "Pulse radiolysis study on the estimation of radiolytic yields of water decomposition products in high-temperature and supercritical water: Use of methyl viologen as a scavenger", *J. Phys. Chem. A*, Vol. 108, 2004, 8287-8295.
- [7] M. Lin, Y. Katsumura, H. He, Y. Muroya, Z. Han, T. Miyazaki, and H. Kudo, "Pulse radiolysis of 4,4'-bipyridyl aqueous solutions at elevated temperatures: Spectral changes and reaction kinetics up to 400 °C", *J. Phys. Chem. A*, Vol. 109, 2005, pp. 2847-2854.
- [8] S. Torquato, "Diffusion and reaction among traps: Some theoretical and simulation results", *J. Stat. Phys.*, Vol. 65, 1991, pp.1173-1206.
- [9] G.R. Freeman, "Basics of radiation chemistry", in *The Study of Fast Processes and Transient Species by Electron Pulse Radiolysis*, J.H. Baxendale and F. Busi (Eds.), D. Reidel Publishing Co., Dordrecht, Holland, 1982, pp.19-34.
- [10] S. Sanguanmith, "Low-linear energy transfer radiolysis of liquid water at elevated temperatures up to 350 °C: Monte-Carlo simulations", M.Sc. Thesis, Université de Sherbrooke, Sherbrooke (Québec), 2012.
- [11] J.W.T. Spinks and R.J. Woods, *An Introduction to Radiation Chemistry*, 3rd ed., Wiley, New York, 1990.
- J. Meesungnoen, D. Guzonas, and J.-P. Jay-Gerin, "Radiolysis of supercritical water at 400 °C and liquid-like densities near 0.5 g/cm³. A Monte Carlo calculation", *Can. J. Chem.*, Vol. 88, 2010, pp. 646-653.
- [13] A.J. Elliot and D.M. Bartels, "The reaction set, rate constants and g-values for the simulation of the radiolysis of light water over the range 20° to 350°C based on information available in 2008", Report AECL No. 153-127160-450-001, Atomic Energy of Canada Ltd., Chalk River, Ontario, 2009.
- [14] S. Sanguanmith, Y. Muroya, J. Meesungnoen, M. Lin, Y. Katsumura, L. Mirsaleh Kohan, D.A. Guzonas, C.R. Stuart, and J.-P. Jay-Gerin, "Low-linear energy transfer radiolysis of liquid water at elevated temperatures up to 350 °C: Monte-Carlo simulations", *Chem. Phys. Lett.*, Vol. 508, 2011, pp. 224-230.
- [15] S. Sanguanmith, Y. Muroya, T. Tippayamontri, J. Meesungnoen, M. Lin, Y. Katsumura, and J.-P. Jay-Gerin, "Temperature dependence of the Fricke dosimeter and spur expansion time in the low-LET high-temperature radiolysis of water up to 350 °C: A Monte-Carlo simulation study", *Phys. Chem. Chem. Phys.*, Vol. 13, 2011, pp.10690-10698.
- [16] S. Sanguanmith, J. Meesungnoen, and J.-P. Jay-Gerin, "Temperature dependence of the spur expansion time in the low-LET radiolysis of water over the range 25 to 350 °C", in <u>Proceedings of the 3rd China-Canada Joint Workshop on Supercritical Water-Cooled Reactors</u>, Xi'an, Shaanxi, China, 18-20 April 2012. Paper No. 12058.
- [17] H.E. Sims, "Yields of radiolysis products from γ-irradiated supercritical water. A reanalysis data by W.G. Burns and W.R. Marsh", *Radiat. Phys. Chem.*, Vol.**75**, 2006, pp.1047-1050.
- [18] Y. Muroya, M. Lin, V. de Waele, Y. Hatano, Y. Katsumura, and M. Mostafavi, "First observation of picosecond kinetics of hydrated electrons in supercritical water", *J. Phys. Chem. Lett.*, Vol. 1, 2010, pp. 331-335; Y. Muroya, unpublished data, 2010.